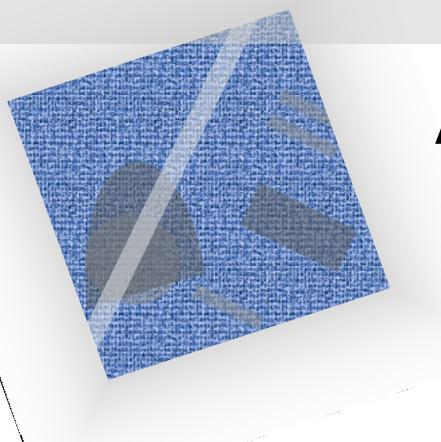
Heat



A Form of Energy

Temperature and Heat

- Kinetic energy is the energy of motion
 - Temperature is the measure of the average kinetic energy of an object

Question

 How is kinetic energy related to heat production?

Thermometer

- A instrument used to measure temperature
- Thermometers commonly have alcohol (with dye) or mercury Digital thermometers have replaced older ones

Celsius Scale

- Celsius is the metric scale for measuring temperature
 - Water freezes at 0°C and boils at 100°C

Kelvin scale

 The Kelvin scale is a metric temperature scale measured in Kelvin units (K)

Formula (273+°C)= Kelvin

Questions

- What is the formula for converting a Celsius temperature to a Kelvin temperature?
- What is the boiling point of water on the Kelvin scale?
- What is the freezing point of water on the Kelvin scale?

Absolute zero

 The temperature in which all molecular motion stops (0 K)

Questions

- Describe absolute zero.What is absolute zero
 - What is absolute zero on the Celsius scale?

Measuring Heat

- Increase in temperature
- **Addition of heat**
- A decrease in temperature
 - Removal of heat

Calories

- Unit for measuring heat
 The amount of heat
 needed to raise 1 gram
 - needed to raise 1 gram of water one degree Celsius

Temperature

- Joule is another unit for measuring heat
- Mass and type of substance determine the amount of temperature change

Specific Heat

- The ability of a substance to absorb heat energy (specific heat)
 - Different substances absorb heat at different rates
 - The greater the mass of the object the more heat is absorbed

What is a calorimeter?

 Device used to measure the heat given off during chemical reactions

Questions

- How can heat be measured?
- What is the unit used to measure heat?What is specific heat?

Heat and Phase Changes

- A phase change is a physical change that requires a change in heat energy
 - Addition or removal of HEAT

Questions

 What is freezing point, melting point, and boiling point?

Heat expansion

The expansion of a substance due to heat
Most solids, liquids, and gases expand as they are heated

Expansion of water

Between 4°C and 0°C, water EXPANDS as it cools and turns into a solid

Thermostat

 A thermostat is a device that controls the temperature The switch of a thermostat is a bimetallic strip

Bimetallic Strip

- Two different metals that are bound together
- They expand at different rates when heated
 - Used as a switch in a thermostat

Internal energy

The energy within a substance

Questions

- What is internal energy?
 - What is a thermostat?
- Describe the switch of a thermostat.
 - What happens to most substances when heat is added?

Molecules and Motion

- The motion of molecules produces heat
 - The more motion, the more heat is generated

Heat Transfer

 The movement of heat from a warmer object to a colder one

Forms of heat transfer

- Three forms of heat transfer:
- ConductionConvectionRadiation

Conduction

Conduction involves the transfer of heat through direct contact **Heat conductors** conduct heat well, insulators do not

Convection

 Takes place in liquids and gases as molecules move in currents
 Heat rises and cold settles to the bottom

Radiation

- Heat is transferred through space
- Energy from the sun being transferred to the Earth

Questions

- What are the three types of heat transfer?
- How is conduction different from radiation?

What type of heat transfer is involved?

- Heating a room with a fireplace
- Egg cooking in a frying pan
 - Roof of a house becoming hot

What type of heat transfer?

- Warm air mass bringing a change in the weather
 - Wire getting hot from an electric appliance